Phytochemistry, 1975, Vol. 14, p. 2717. Pergamon Press. Printed in England.

# ALKALOIDS OF *URGINEA ALTISSIMA* AND THEIR ANTIMICROBIAL ACTIVITY AGAINST *PHYTOPHTHORA CAPSICI*

# MASAKAZU MIYAKADO, TOSHIRO KATO, NOBUO OHNO and KOICHI KOSHIMIZU\*

The Pesticide Department, Institute for Biological Science, Sumitomo Chemical Company, Takarazuka, Hyogo, Japan, \*Department of Food Science and Technology, Faculty of Agriculture, Kyoto University, Kyoto, Japan

(Received 27 May 1975)

Key Word Index—Urginea altissima, Lilliaceae, alkaloids, lycorine, acetylcaranine, antimicrobial activity; Phytophthora capsici

Plant and source. Urginea altissima Bak. Plant material was collected at northern part of the Lake Rudolf, Ethiopia.

Present work. 950 g of fresh bulb tissues were extracted with MeOH. The extract showed considerable antimicrobial activity against *Phytophthora capsici*. This activity was located in the water-soluble part of the extract which (chromatographed over alumina) gave lycorine [1], (1), mp 273–278°,  $[\alpha]_{D}^{22^{\circ}}$  – 125 3° (EtOH), (IR, NMR) as an active component. Lycorine showed protective activity against blight (*P. capsici*) of tomato plants, when an aq. soln (500  $\mu$ g/ml) was sprayed on the leaves. The HCl salt exhibited a slight enhancement in activity, but diacetyllycorine, dihydrolycorine and 2-acetyllycorine possessed lower activity than that of lycorine itself.

The CHCl<sub>3</sub> soluble part of MeOH extract (chromatographed over Si gel, TLC) gave the alkaloid, acetylcaranine (2),  $C_{18}H_{19}NO_4$ ,  $M^+$  313, mp 173–174°,  $\lceil \alpha \rceil_D^{22^\circ} - 80.5^\circ$  (CHCl<sub>3</sub>),  $\nu_{max}$ 

1735 cm<sup>-1</sup> (acetate),  $\lambda_{\text{max}}^{\text{McOH}}$  290 nm ( $\epsilon$ , 2900), NMR (CDCl<sub>3</sub>, TMS):  $\delta$  6·72, 6·57 (each 1H, s, for H-11, H-8), 5·89 (2H, s, for H-12 methylene dioxy), 5·85† (1H, br s, for H-1), 5·40 (1H, br s, for H-3), 4·15, 3·53 (each 1H, d, J 15 Hz, for H-7 $\alpha$ , H-7 $\beta$  and 1·91 (3H, s, for acetyl-Me). (2) was identical with an authentic specimen (derived from caranine by acetylation; mmp, IR, NMR). This is the first report of (2) as a natural product.

(1) Lycorine R = OH R'= H

(2) Acetylcaranine R = H R' = Ac

Acknowledgements—The authors thank Dr K Kotera and Dr K Tori. Shionogi Research Laboratory, Japan for authentic samples of lycorine, caranine and many helpful discussions We also thank Mrs M Fukui for plant collections

### REFERENCES

1 Kotera, K, Hamada, Y, Tori, K, Aono, K and Kuriyama, K (1966) Tetrahedron Letters, 2009 (and references cited therein)

Phytochemistry 1975, Vol 14, pp 2717-2718 Pergamon Press Printed in England

# TETRANORTRITERPENOIDS FROM THE HEARTWOOD OF CARAPA GUIANENSIS

## GUIDO B. MARCELLE and BALDWIN S. MOOTOO

Department of Chemistry, The University of the West Indies, St Augustine, Trinidad, West Indies

(Received 20 May 1975)

Key Word Index—Carapa guianensis, Mehaceae, 6α11β-diacetoxygedunin; 6α-acetoxygedunin

Plant. Carapa guianensis Aubl. (Meliaceae). Previous work. The seeds of this plant have yielded

several tetranortriterpenoids [1,2]. The heartwood, obtained from Trinidad, West Indies, has

<sup>+</sup>The extraordinary low field shift of H-1 appears to be benzene ring anisotropy and van der Waals' effect between H-1, H-11 protons, the latter was evident on a NOE experiment

shown 11 $\beta$ -acetoxygedunin and  $6\alpha$ ,11 $\beta$ -diacetoxygedunin [3]

Present work. We re-investigated the heartwood of this plant, obtained locally, and were able to isolate  $6\alpha,11\beta$ -diacetoxygedunin (1) and  $6\alpha$ -acetoxygedunin (2) The latter has not been previously reported from the heartwood

The CHCl<sub>3</sub>-soluble fraction of the EtAc extract of the ground heartwood yielded by PLC two crystalline compounds.  $6\alpha.11\beta$ -diacetoxygedunin (1) and  $6\alpha$ -acetoxygedunin (2).

6α,11β-diacetoxygedunin (1) crystallized from MeOH mp 195-196′, 247-248°;  $\lambda_{max}$  222 nm ( $\epsilon$  11500);  $\nu_{max}$ 1740-1750, 1675 and 874 cm<sup>-1</sup> On mild alkaline hydrolysis the compound yielded the expected triol 6α,7α,11β-trihydroxy-7-deacetoxygedunin (3) which crystallized from EtOAc mp 266–268′;  $\lambda_{max}$  218 nm ( $\epsilon$  10700); IR  $\nu_{max}$  3470, 1740, 1650 and 870 cm<sup>-1</sup> Acetylation of the above triol (3) with Ac<sub>2</sub>O-pyridine at room temp. gave the diacetate 7-hydroxy-6α,11β-diacetoxy-7-deacetoxygedunin (4) as crystals from EtOAc mp 289-291°,  $\lambda_{max}$  218 nm ( $\epsilon$  10600);  $\nu_{max}$  3470, 1735, 1675 and 874 cm<sup>-1</sup>.

A comparison of the NMR spectrum of this diacetate (4) with that of the naturally occurring triacetate  $6\alpha.11\beta$ -diacetoxygedunin (1) confirmed that it was the product which under these conditions resulted from the  $C_7$  hydroxyl resisting acetylation Thus, there were 2 acetoxy methyl signals in the NMR ( $\delta$  2.12, 2.19). In addition there were 2 protons (CHOAc) centred at  $\delta$ 5.76 and  $\delta 5.34$  which corresponded in both shape and position to those assigned to the  $C_{11}$  and  $C_6$  protons respectively of the naturally occurring triacetate [3]. A broad signal in the spectrum of the diacetate (4) at  $\delta$ 351 replaced the doublet at  $\delta$ 4.89 which had been assigned to the  $C_7$  proton of  $6\alpha$ ,  $11\beta$ -diacetoxygedunin On shaking with  $D_2O$ the signal sharpened to a doublet (J 3 Hz) and must be the proton on the hydroxyl bearing  $C_7$ 

6α-Acetoxygedunin (2) crystallized from EtOAc-light petroleum mp 270-273 : λ<sub>max</sub>

220 nm ( $\epsilon$  10000);  $v_{\text{max}}$  1760, 1740, 1670 and 870 cm<sup>-1</sup>. The signals in the NMR spectrum corresponded to those reported previously [2].

#### EXPERIMENTAL

IR spectra were in Nujol and UV in EtOH TLC and PLC were on Si gel (Merck 60 PF<sub>254-366</sub>) Mp's are uncorrected Light petrol was bp 60-80

Extraction and isolation Ground heartwood (4kg) was extracted with EtOAc (191) for 2 days. The reddish-brown extract was evaporated to dryness and the CHCl<sub>3</sub>-soluble residue (17 g) washed with light petiol leaving an insoluble gum (13 6g) PLC (18 g) on a large plate  $(40 \times 60 \,\mathrm{cm}, 2 \,\mathrm{mm}$  thick) with light petrol-acetone (3.1) gave 2 main bands. The band of higher  $R_J$  gave, on repeated PLC, crystalline 62-acetoxygedunin (333 mg) and that of lower  $R_J$  gave  $(6x, 11\beta)$ -diacetoxygedunin (157 mg)

62,11 $\beta$ -Diacetoxygedium; (1): mp. 195-196°, 247-248°, M° 598. (Found, C. 63.9), H., 6.5. Calc. for  $C_{52}H_{59}O_{19}$ ,  $C_{11}$ 64.2., H. 6.4%.)

6α 7α,11β-Trihydroxy-7-deacetoxygedunin (3) Treatment of 6α,11β-diacetoxygedunin (100 mg) with  $5^{\circ}_{\circ}$  KOH-MeOH (5 ml) overnight at R T gave the triol (60 mg) mp 266-268°, M;  $^{+}$  472 (Found: C. 65.9°, H:  $7.0^{\circ}$  C<sub>26</sub>H<sub>32</sub>O<sub>8</sub> expures: C. 66.1: H, 6.8%)

7*τ*-Hvdroxy-6α.11β-diacetoxy-7-deacetoxygedumn (4) was obtained by dissolving the triol (3) in Ac<sub>2</sub>O (2 ml) and pyridine (2 ml) and standing at room temp overnight Work up in the usual way gave the diacetate (95 mg) mp 289 291 , M<sup>4</sup> 556 (Found C. 64 0, H. 65  $C_{30}H_{36}O_{10}$  requires C 64 7, H, 6.5%) NMR (100M Hz, CDCl<sub>3</sub>).  $\delta$  1.22 (3H, s), 1.30 (3H, s), 1.34 (3H, s), 1.43 (6H, s), 2.12 (3H, s), 2.19 (3H, s), 2.68 (1H, d J 5 Hz), 2.73 (1H, d J 12 Hz), 3.51 (1H m. W<sub>1.2</sub> 6 Hz), 3.83 (1H, s), 5.34 (1H, q, J 3.12 Hz), 5.59 (1H, s). 5.76 (1H, m. W<sub>1.2</sub> 21 Hz), 5.92 (1H, d J 10 Hz), 6.30 (1H, m), 7.11 (1H, d, J 10 Hz), 7.38 (2H, m)

62-4cetoxygedunin (2) Mp 270-273 , M $^+$  540 (Found C, 663, H 68, Calc for  $C_{30}H_{36}O_9$  C 666, H, 67° $_o$ ) NMR (60M Hz, CDCl $_3$ )  $\delta$  118 (3H, s), 122 (3H, s), 129 (9H, s), 204 (3H, s), 216 (3H, s), 2:55 (1H, d J 13 Hz), 3:65 (1H s), 494 (1H, d, J 3 Hz), 5:34 (1H, q, J 3 13 Hz), 5:66 (1H, s), 5:99 (1H d, J 10 Hz), 6:41 (1H m), 7:16 (1H d, J 10 Hz), 7:50 (2H, m)

Acknowledgements—The authors are indebted to Dr J Connolly for useful discussions and for recording the MS and 100 M Hz NMR spectrum and to Mr H Séguin for microanalyses

#### REFERENCES.

- Ollis, W. D., Ward A. D., Meirelles de Oliveira, H. and Zelnik, R. (1970) Tetrahedron 26, 1637
- 2 Lavie, D. Levy, E. C. and Zelnik, R. (1972) Bioorganic Chem. 2, 59
- 3 Connolly, J. D., McCrindle, R. Overton, K. H. and Feeney, J. (1966) *Tetrahedron* 22, 891